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bottom filter the solution and mass the remaining solid, if the mass is the same then the solution was saturated, if the mass is lower than it is an unsaturated solution. Model 2 - Solute Dissolved vs. Solute Added The following data refer to an experiment in which a measured mass of solid is added to 10.0 g of 20 °C

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water.

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Container Experiment C (Adding more  
gas) C1 C3C2 Volume = 1 unit Volume =  
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pressure = 1 atm Internal pressure = 1  
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adding 1.0 M HCl to this buffer until a  
total of 101 mL of 1.0M HCl have been  
added. At this point, the pH of the



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solution will be very acidic.

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